

# Forum technologique du CONSOREM

## Abitibi Géosciences 2008

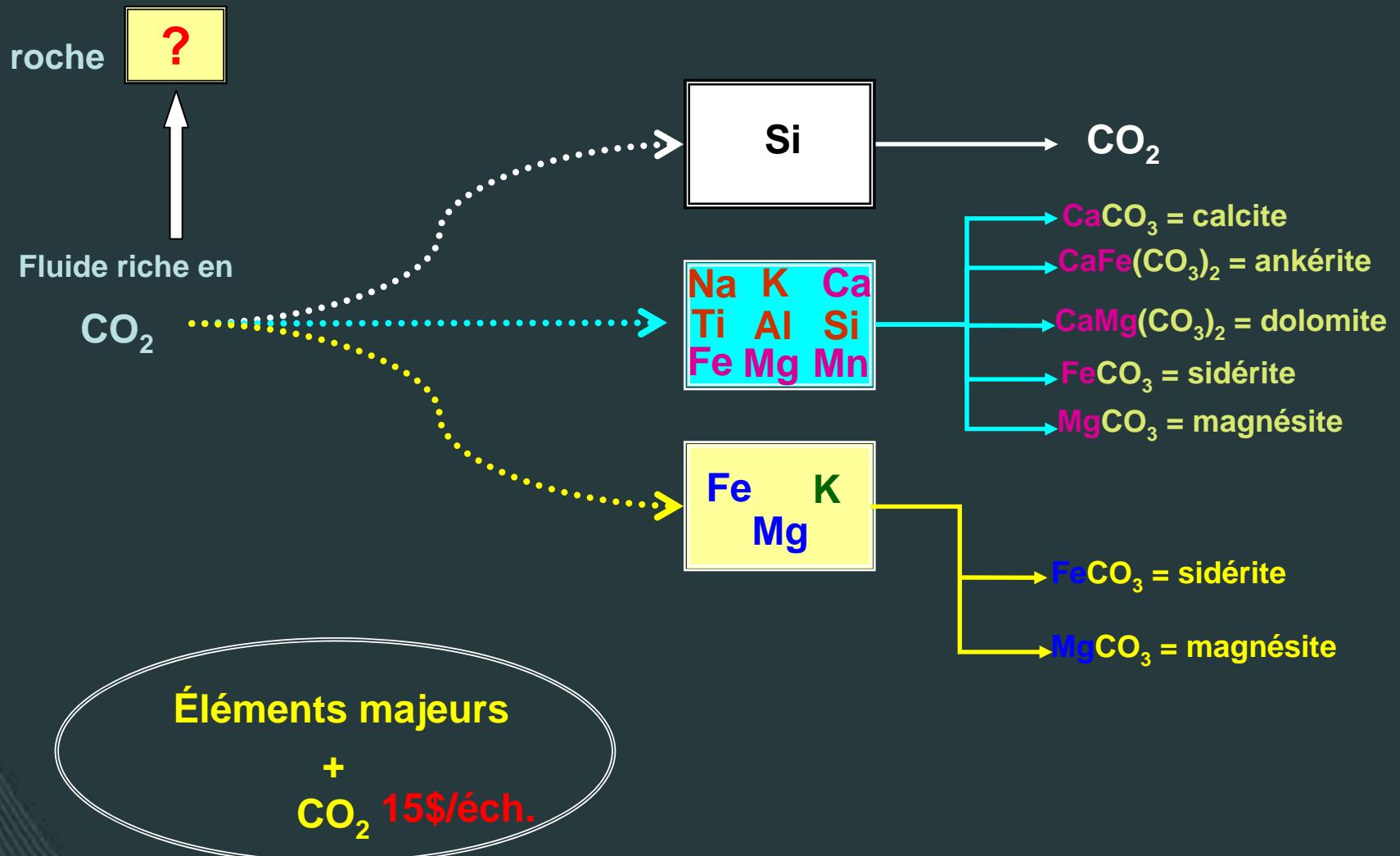
### L'indice de carbonatation: un outil pour l'exploration

Hassan Nabil et Réal Daigneault  
CONSOREM





# Introduction





## Principes de base



CO<sub>2</sub>/CaO molaire = 1 

$\text{CaCO}_3$  (CaO + CO<sub>2</sub>) = calcite

CO<sub>2</sub>/CaO molaire = 2 

$\text{CaFe}(\text{CO}_3)_2$  (CaO + FeO + 2CO<sub>2</sub>) = ankérite



## Indices de carbonatation

### Indice de saturation

- $\text{CO}_2/\text{CaO}+\text{FeO}+\text{MgO}+\text{MnO}$
- La composition des roches a très peu d'influence sur l'indice (attention toutefois aux roches riches en Py-As).

### Indice de discrimination

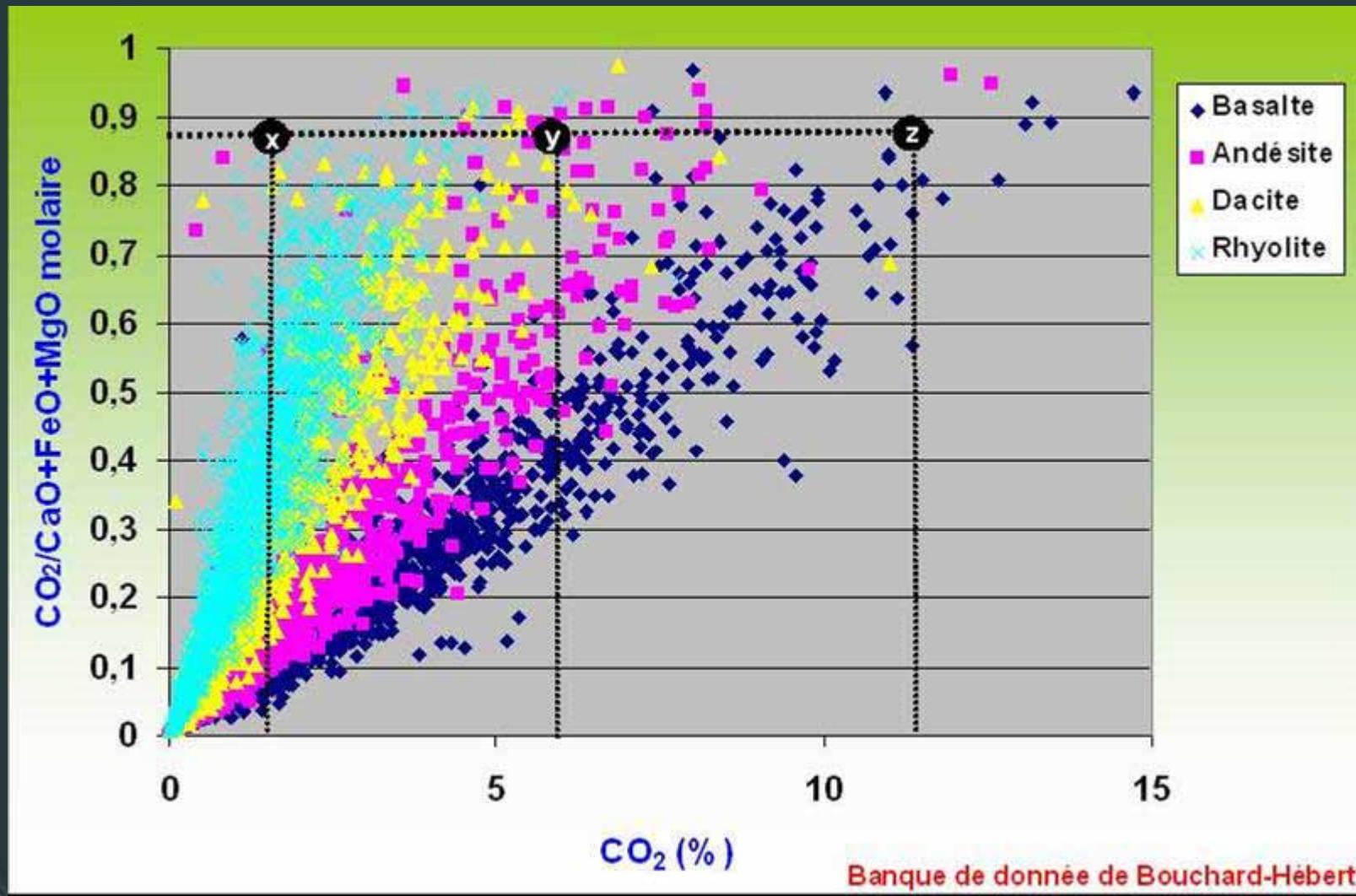
- $\text{CO}_2/\text{CaO}$
- Discriminer les différentes phases de la carbonatation.

$$\text{IS} = \frac{(\% \text{CO}_2 / 44.01)}{((\% \text{MgO} / 40.3) + (\% \text{CaO} / 56.08) + (\% \text{MnO} / 70.94) + (\% \text{Fe}_2\text{O}_3 * (0.8998 / 71.85)))}$$

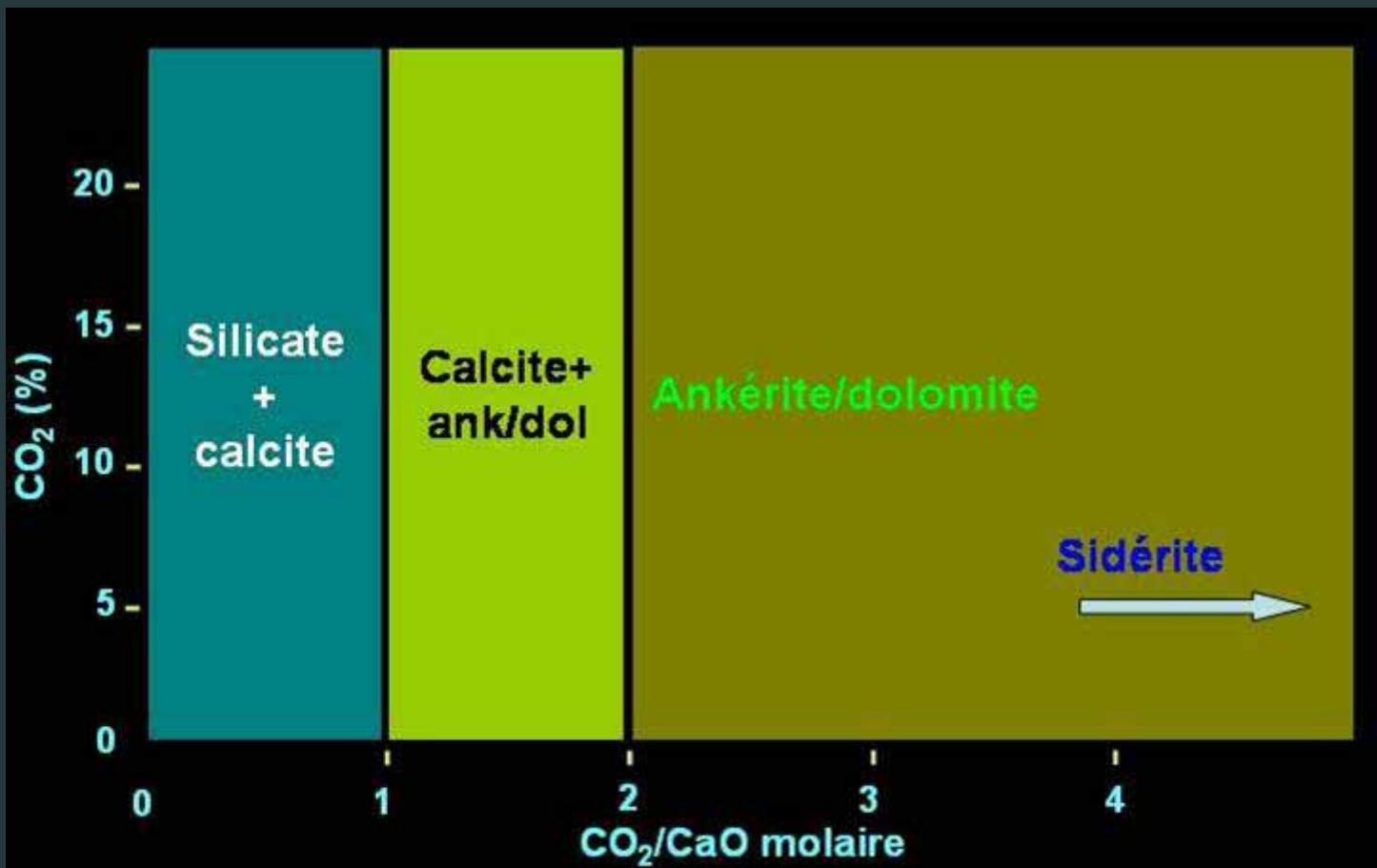
$$\text{ID} = (\% \text{CO}_2 / 44.01) / (\% \text{CaO} / 56.08)$$



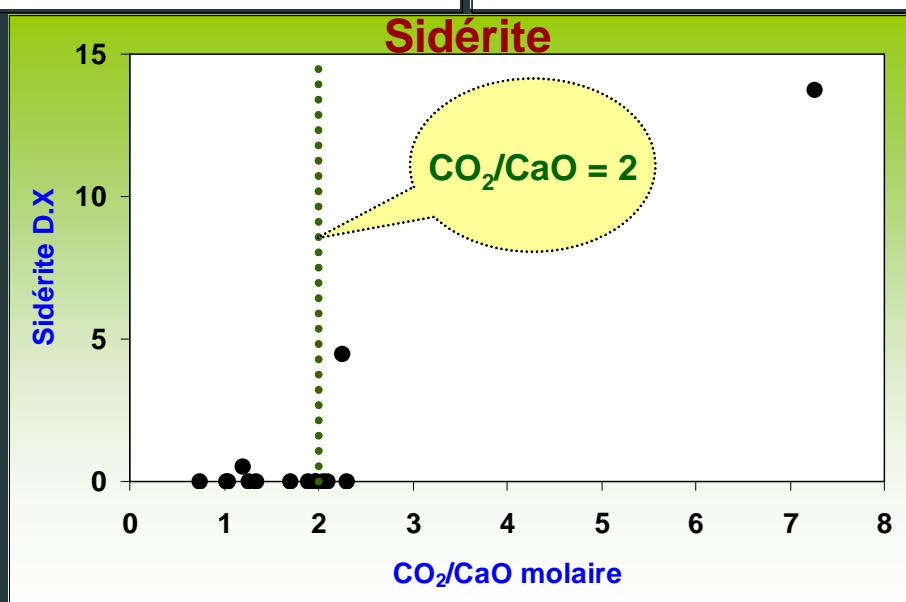
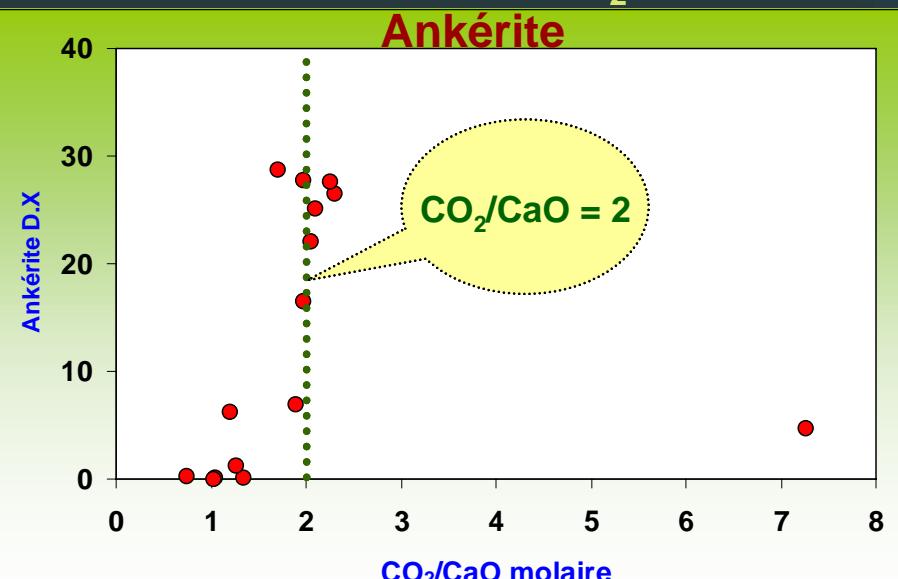
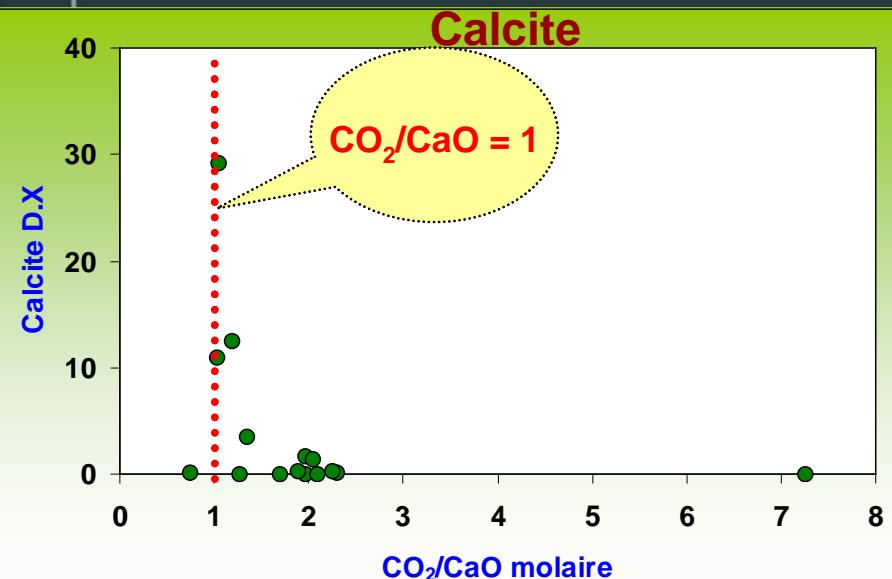
## La teneur en CO<sub>2</sub> ne reflète pas l'intensité de la carbonatation



## CO<sub>2</sub>/CaO molaire

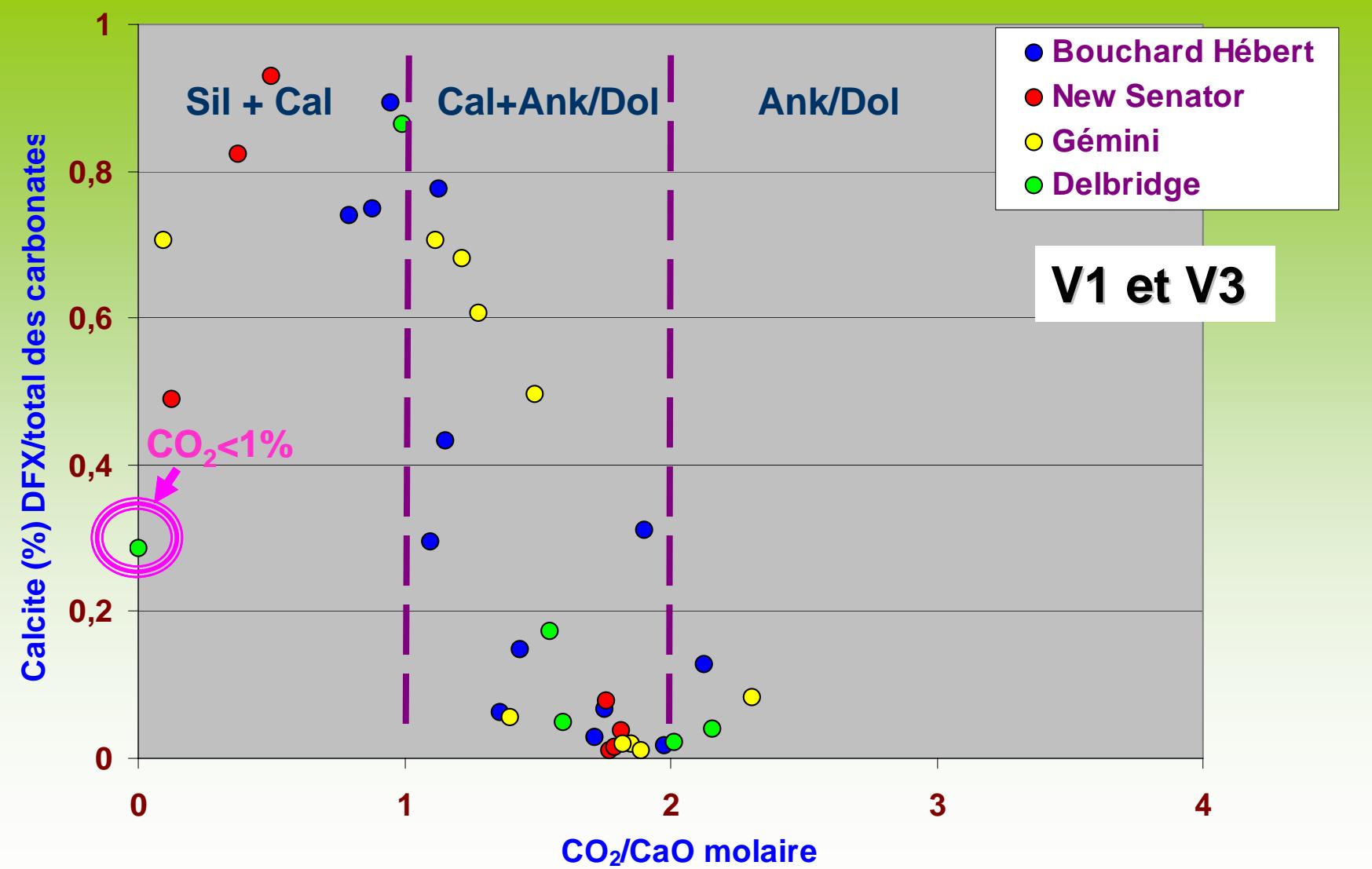


Modifié de Davies et al., 1982

Diffraction X Phases carbonatées vs ratios molaires  $\text{CO}_2/\text{CaO}$ 

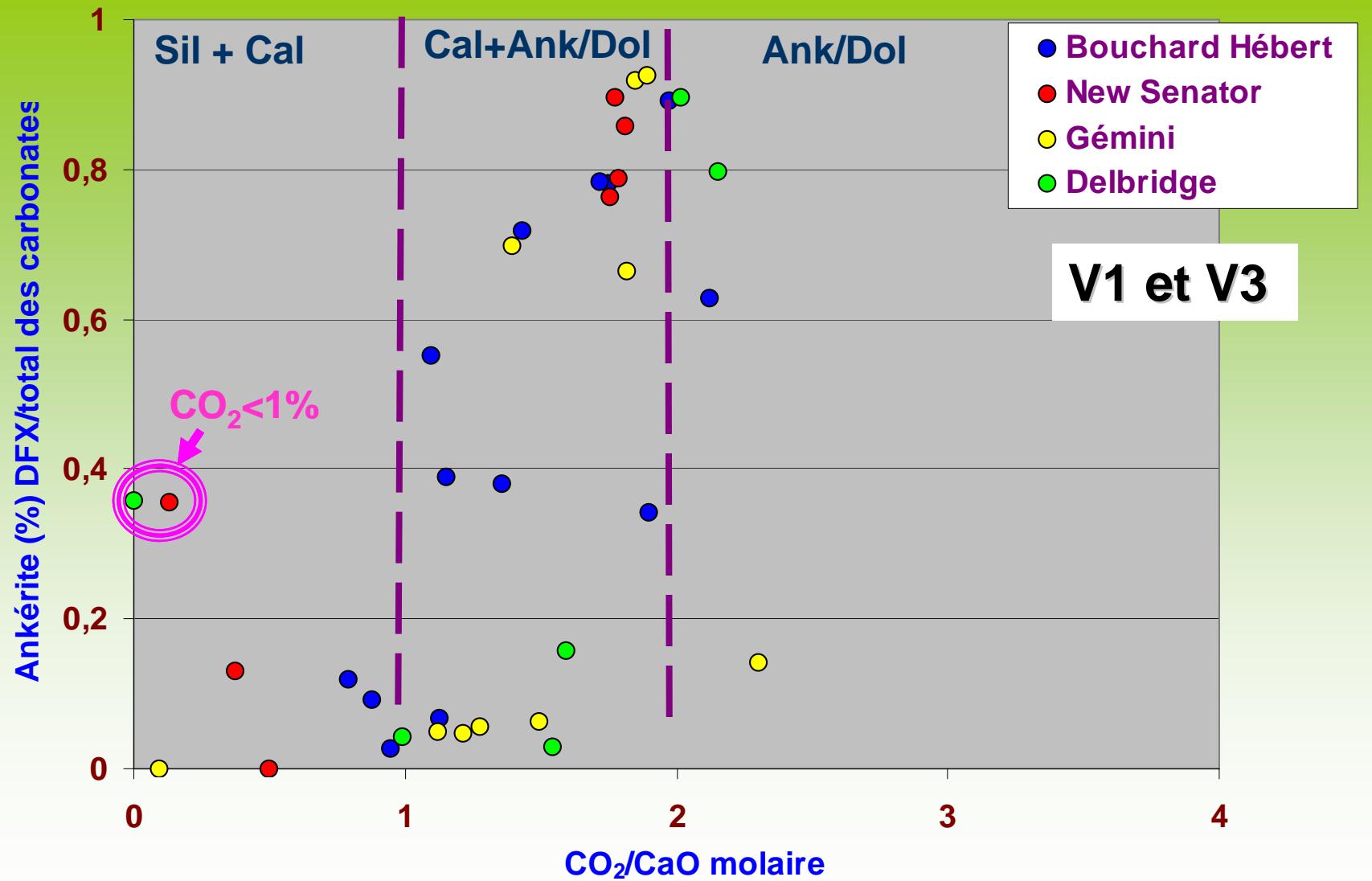


## Abondance Calcite



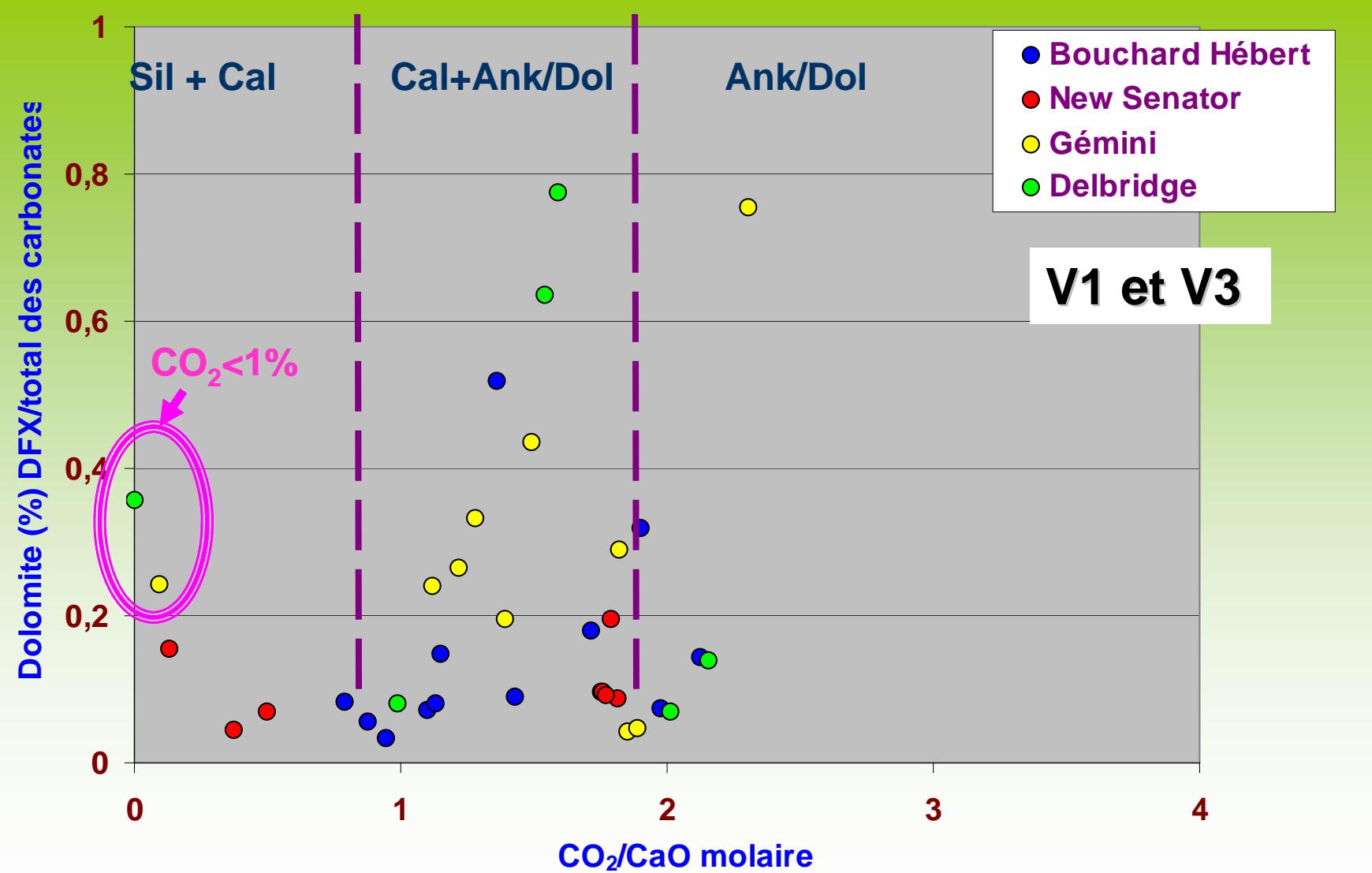


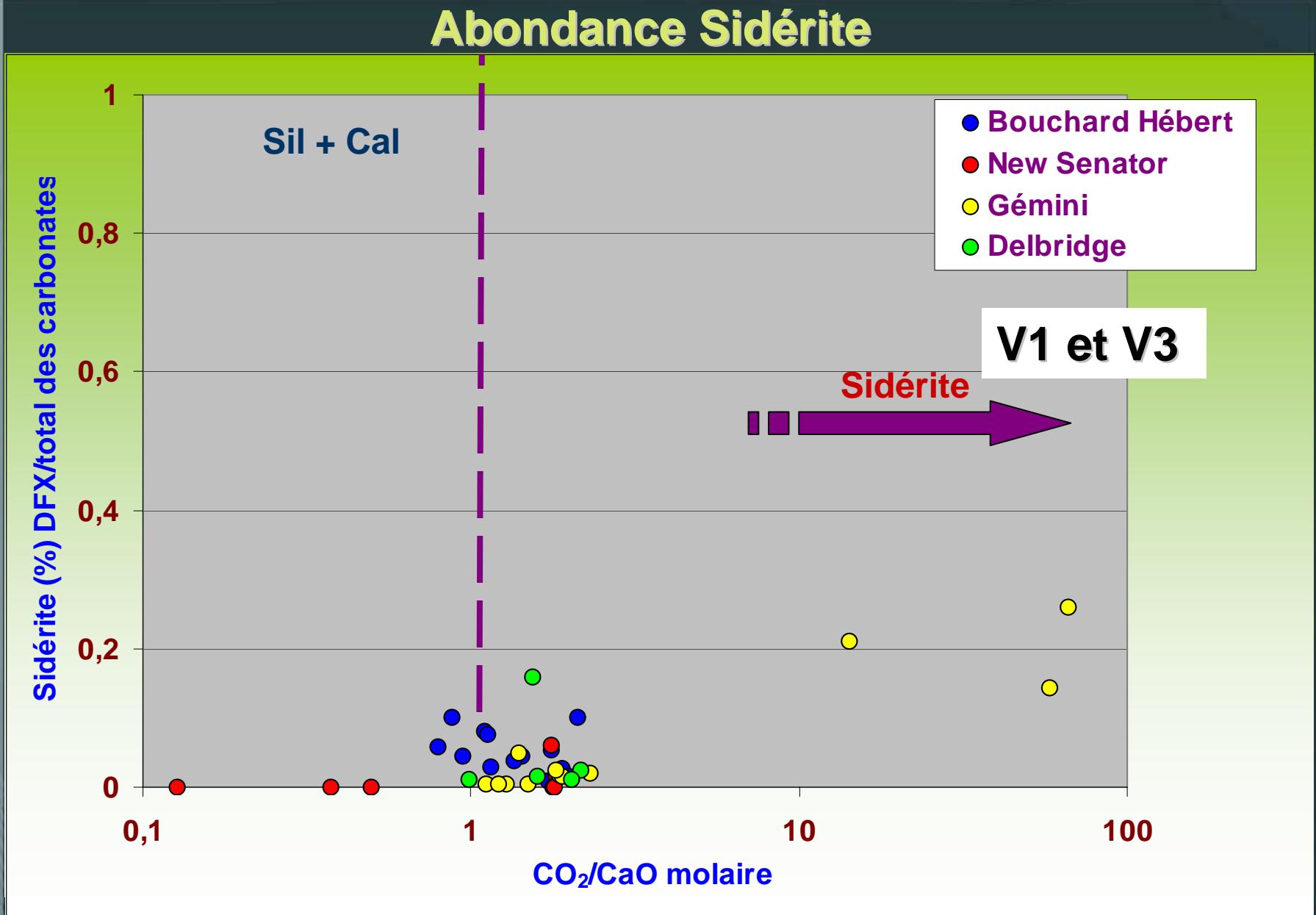
## Abondance Ankérite





## Abondance Dolomite



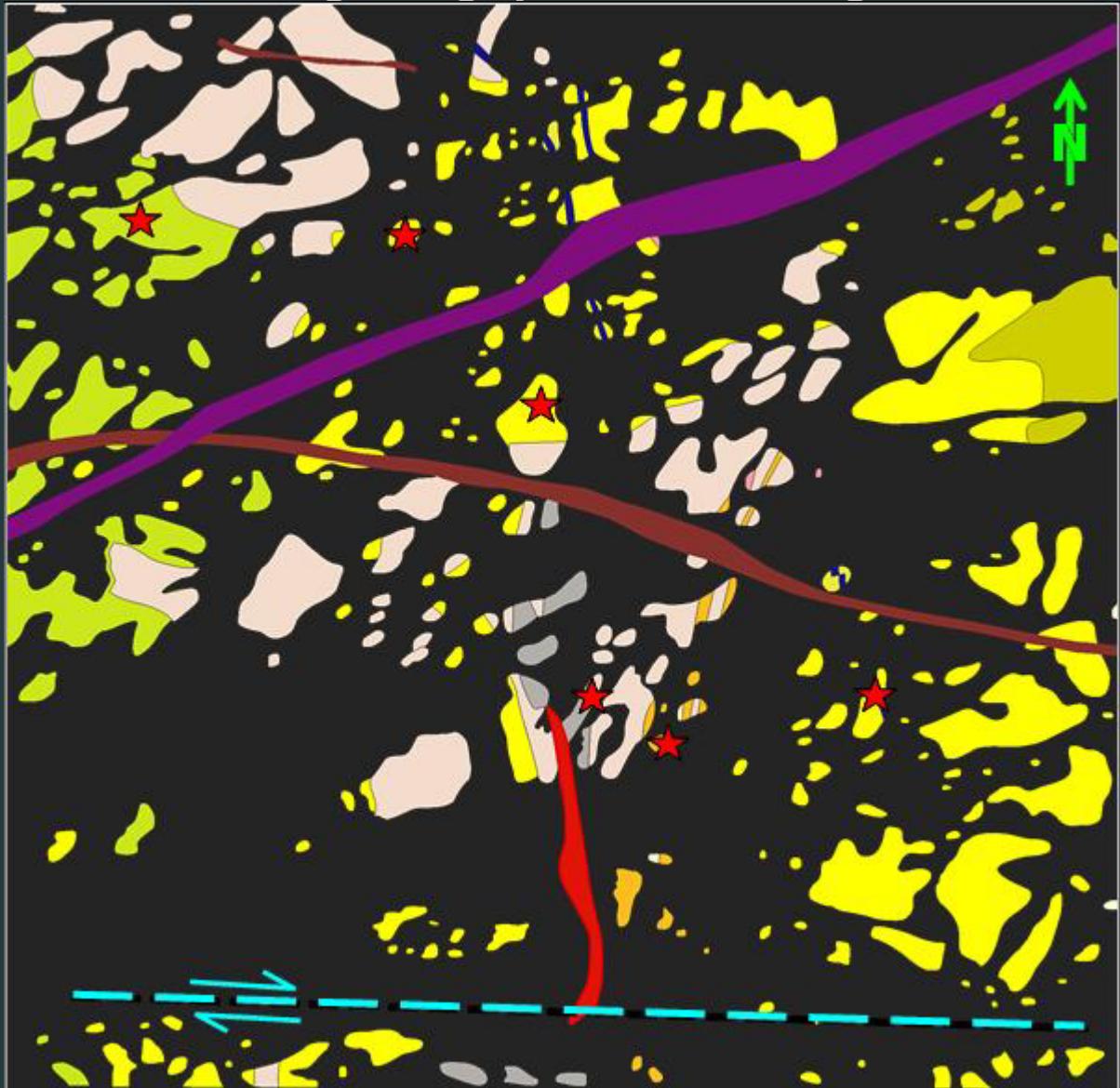




## Carte géologique de Delbridge

- I3B
- I3A
- I3O
- V1B porphyrique
- V1B
- V1D
- V1B bréchifié
- SF.M

120 m





## Exemple d'application

I.S = 0

 $\text{CO}_2/\text{CaO} = 0.1$ 

Calcite: 0.3%

Ankérite: 0.3%

Dolomite: 0.3%

Calcite: 0.6%

Ankérite: 12 %

Dolomite: 2%

I.S = 0.8

 $\text{CO}_2/\text{CaO} = 1.6$ 

Calcite: 0.7%

Dolomite: 12%

Ankérite: 2%

Sidérite: 0.2%

I.S = 0.7

 $\text{CO}_2/\text{CaO} = 1.5$ 

Calcite: 1.5%

Dolomite: 5%

Ankérite: 2%

I.S = 0.5

 $\text{CO}_2/\text{CaO} = 1$ 

Calcite: 20%

Ankérite: 1%

Dolomite: 2%

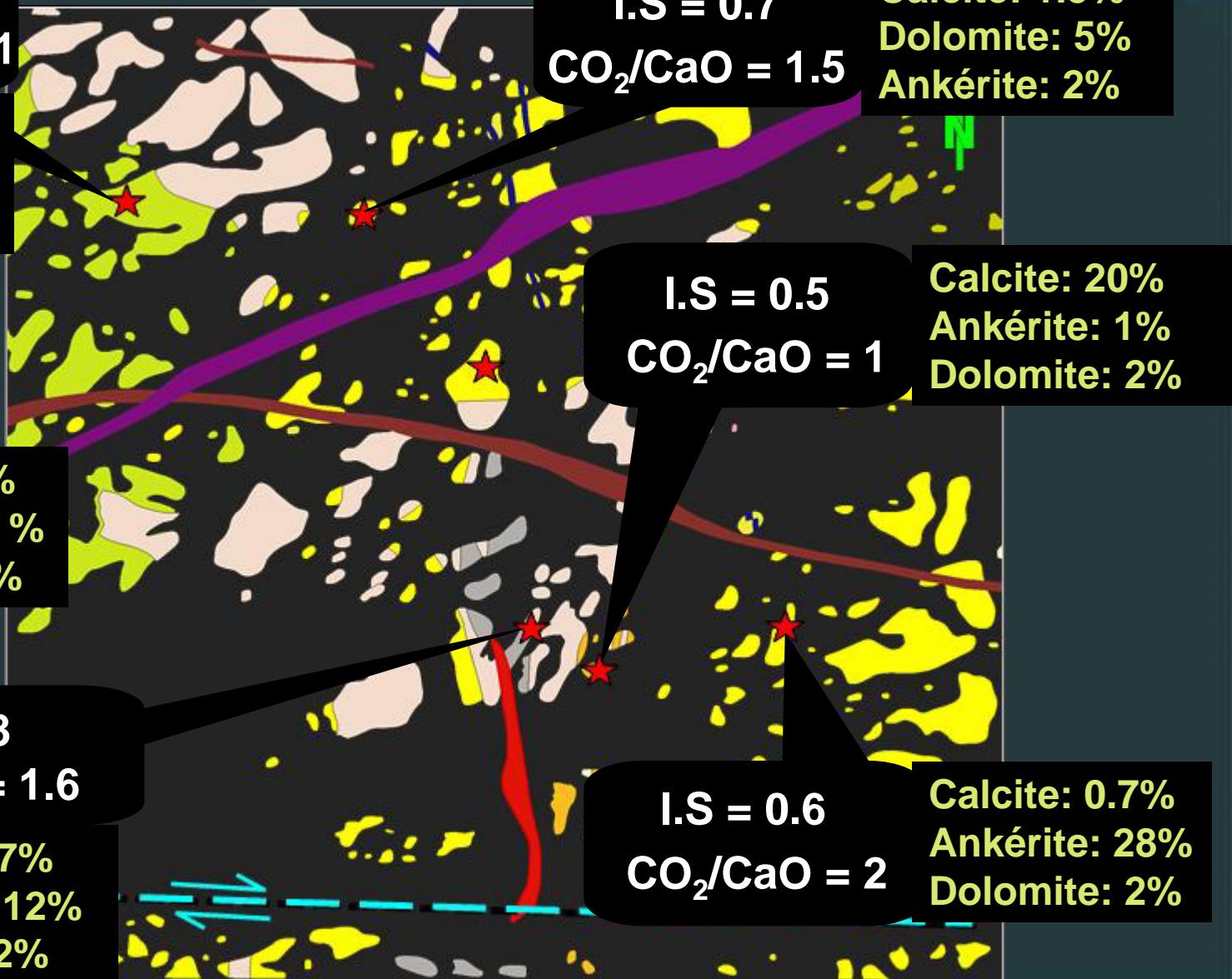
I.S = 0.6

 $\text{CO}_2/\text{CaO} = 2$ 

Calcite: 0.7%

Ankérite: 28%

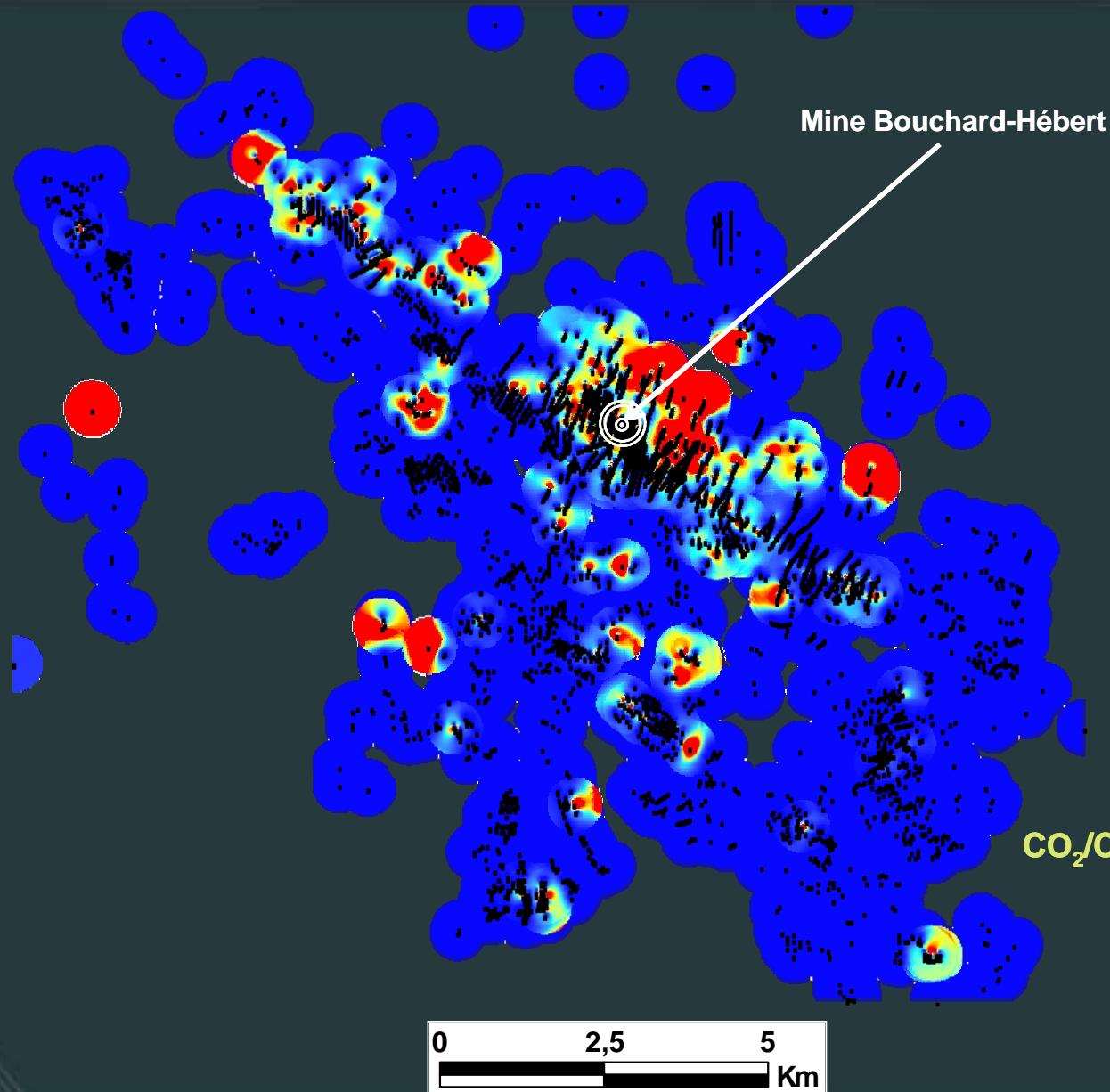
Dolomite: 2%





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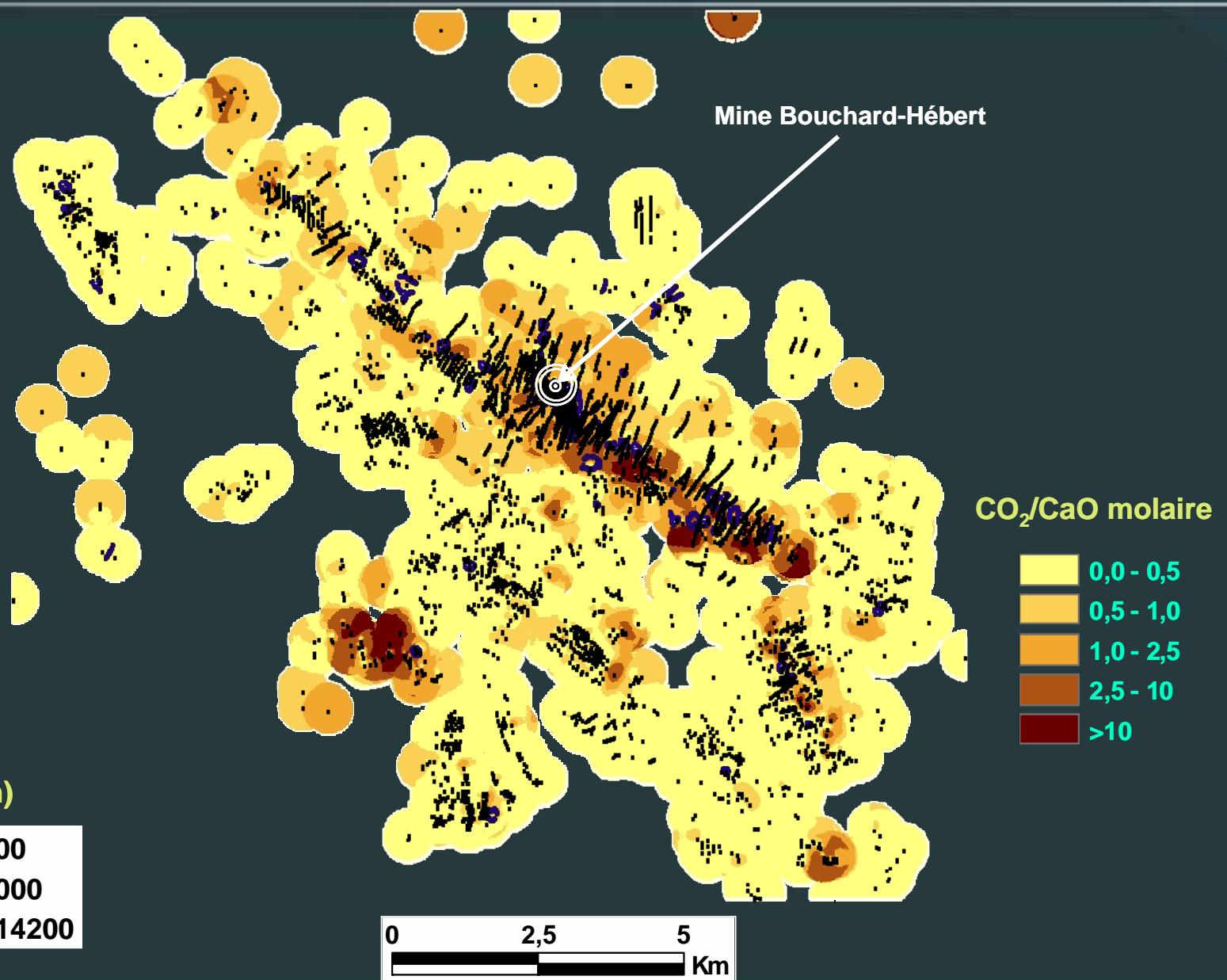
## Indice de saturation

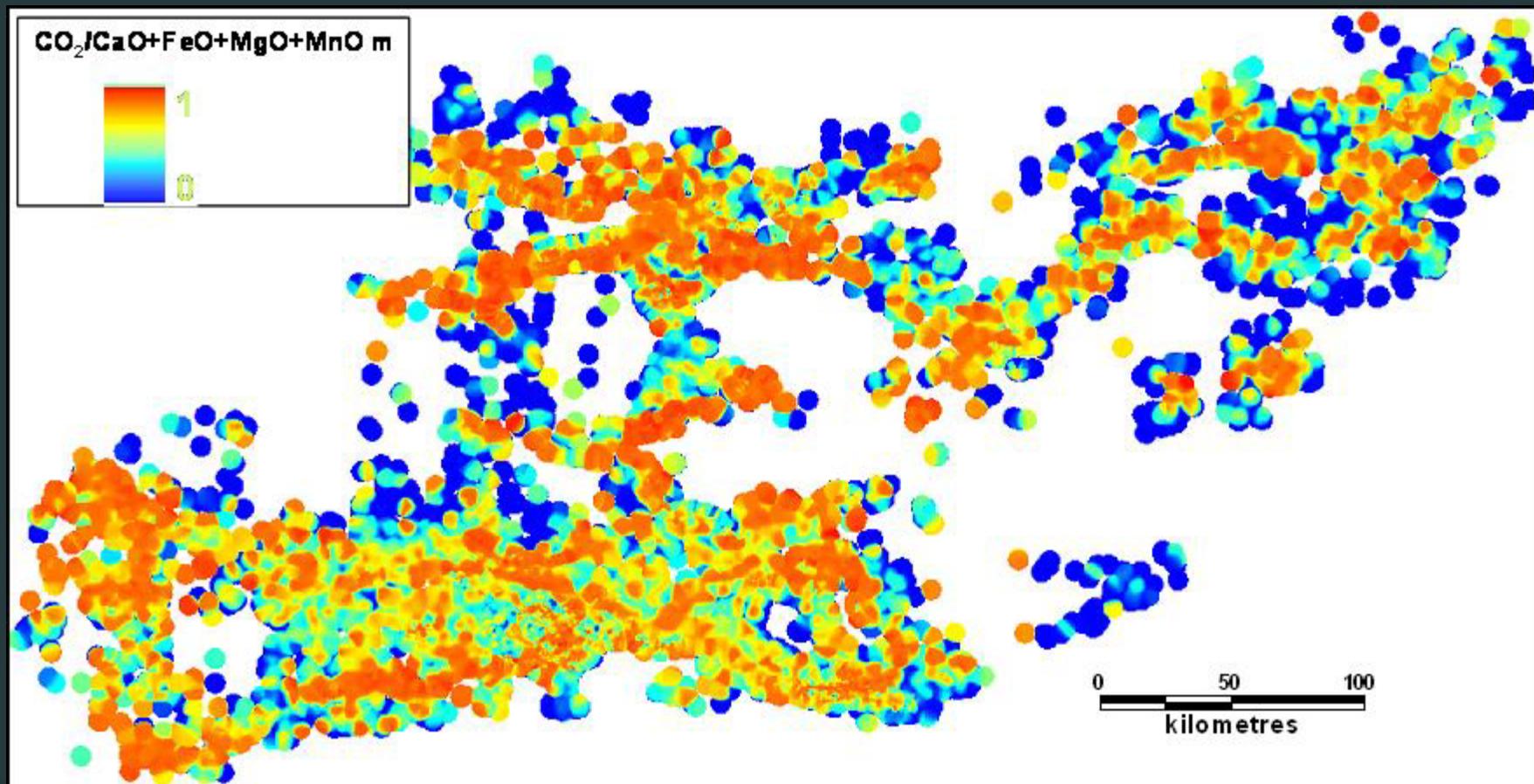


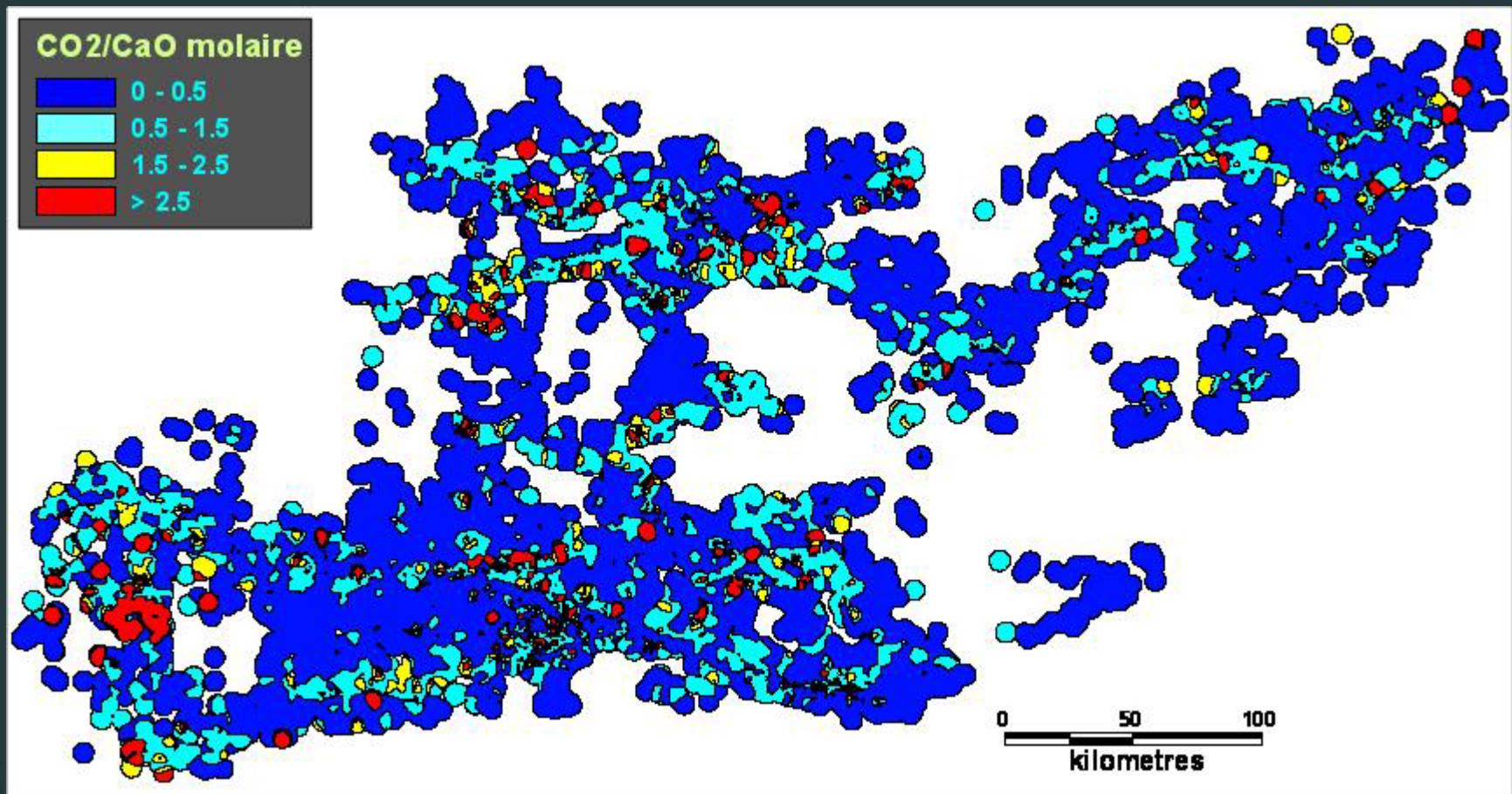


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## Indice de discrimination

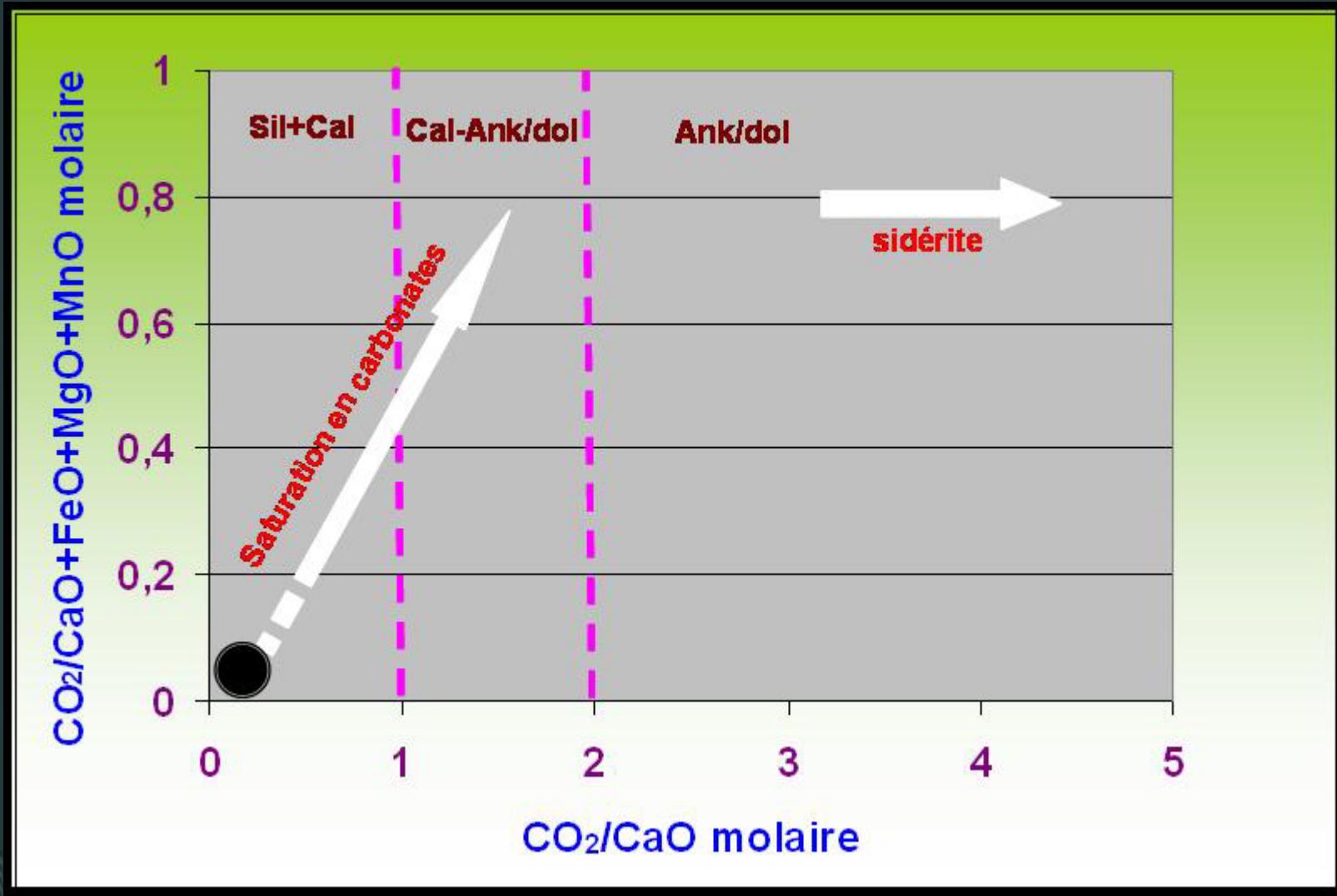


*Indice de saturation*

*Indice de discrimination*

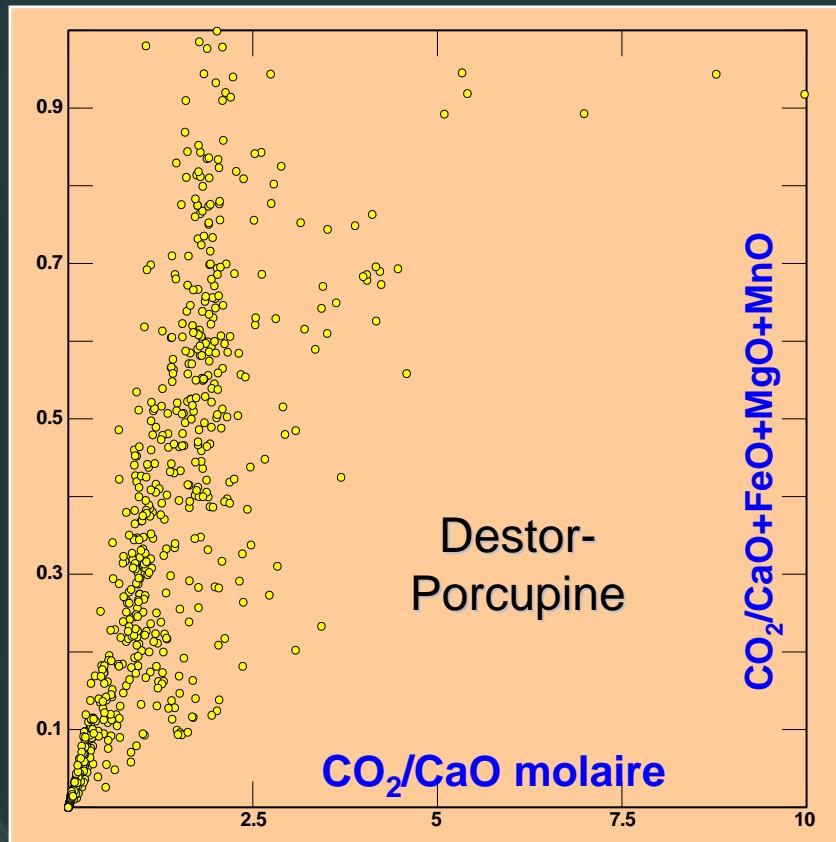


## Diagramme de carbonatation





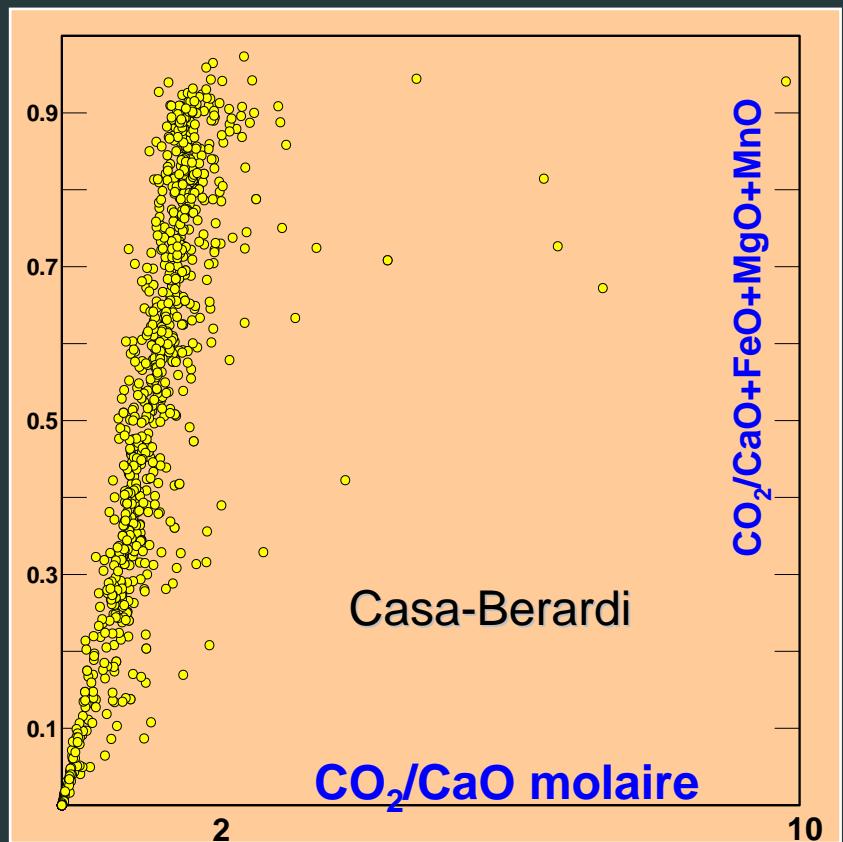
## Exemples pour contextes Or orogénique



Destor-  
Porcupine

$\text{CO}_2/\text{CaO}$  molaire

$\text{CO}_2/\text{CaO} + \text{FeO} + \text{MgO} + \text{MnO}$



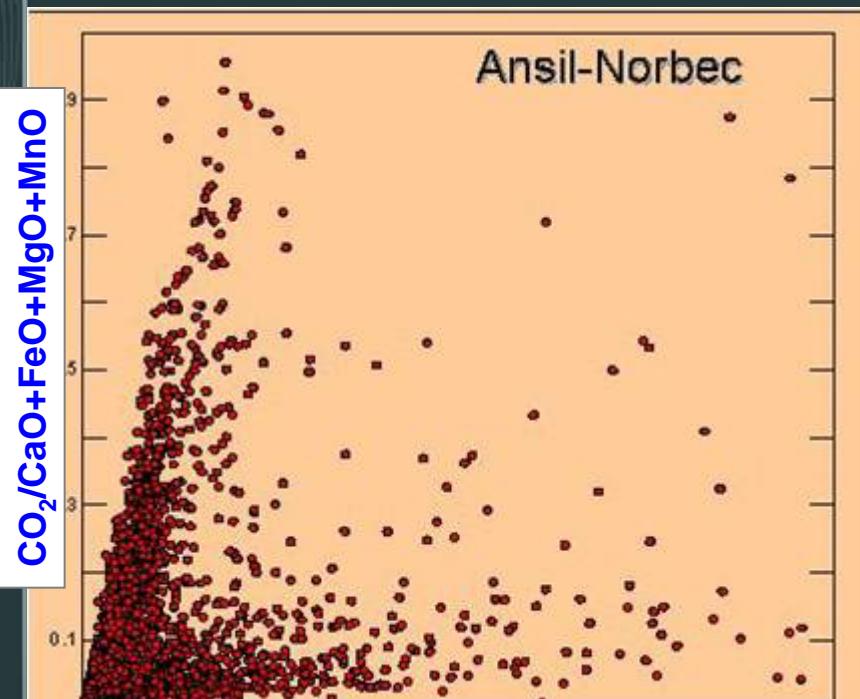
Casa-Berardi

$\text{CO}_2/\text{CaO}$  molaire

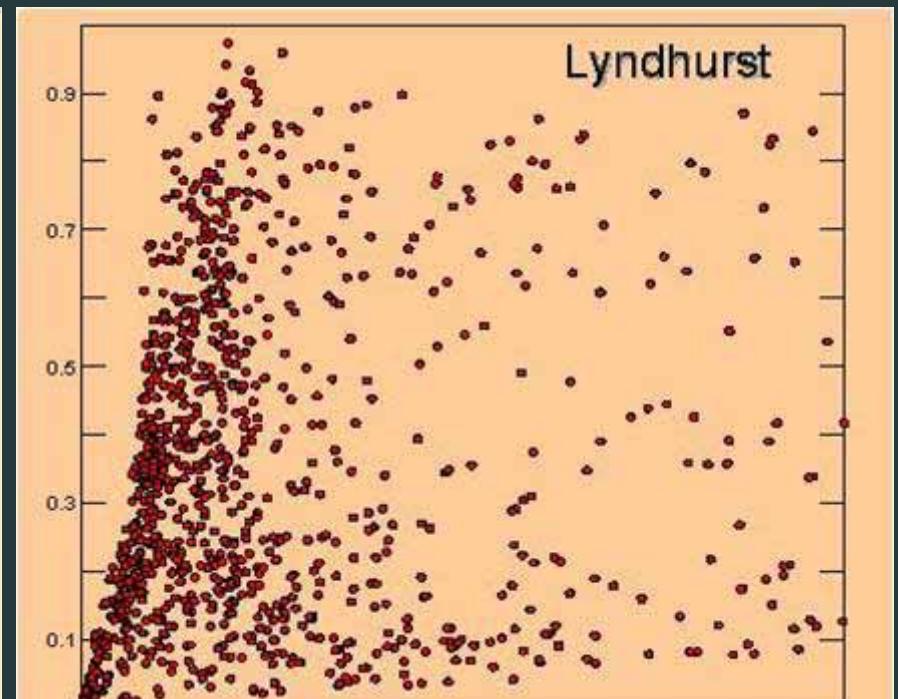
$\text{CO}_2/\text{CaO} + \text{FeO} + \text{MgO} + \text{MnO}$



## Exemples pour contextes VMS

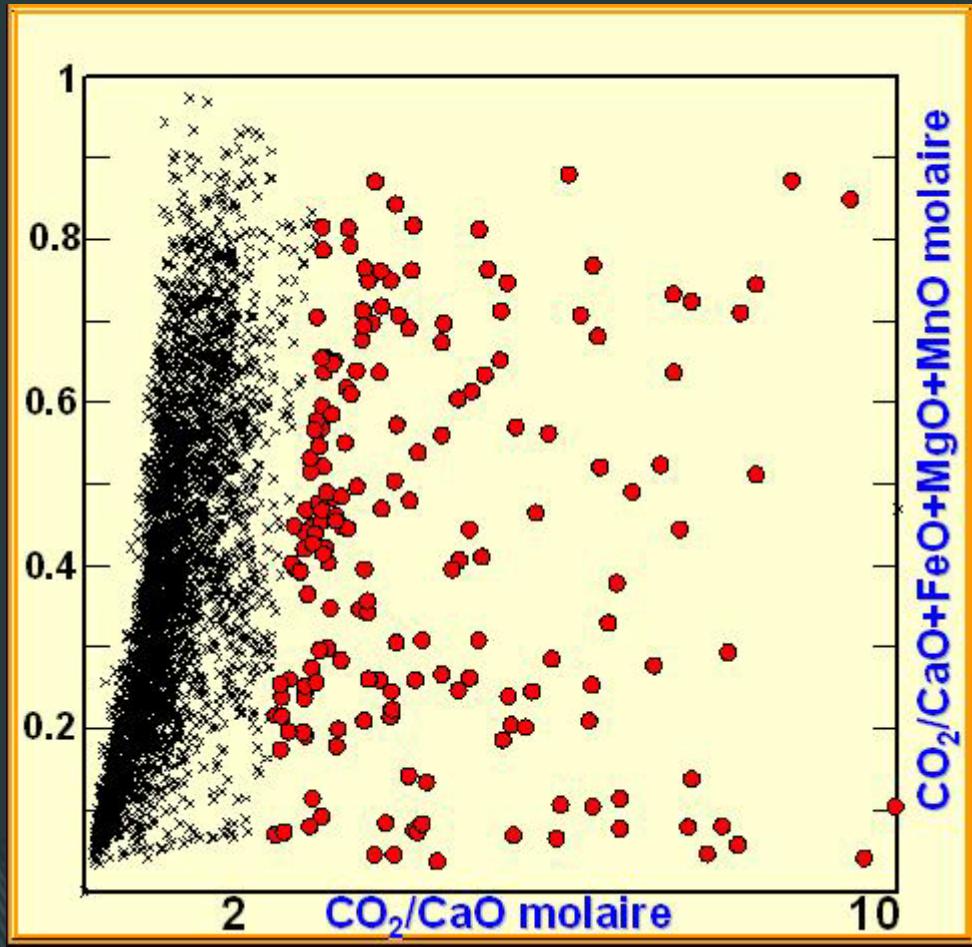


$\text{CO}_2/\text{CaO}$  molaire



$\text{CO}_2/\text{CaO}$  molaire

## Bouchard-Hébert

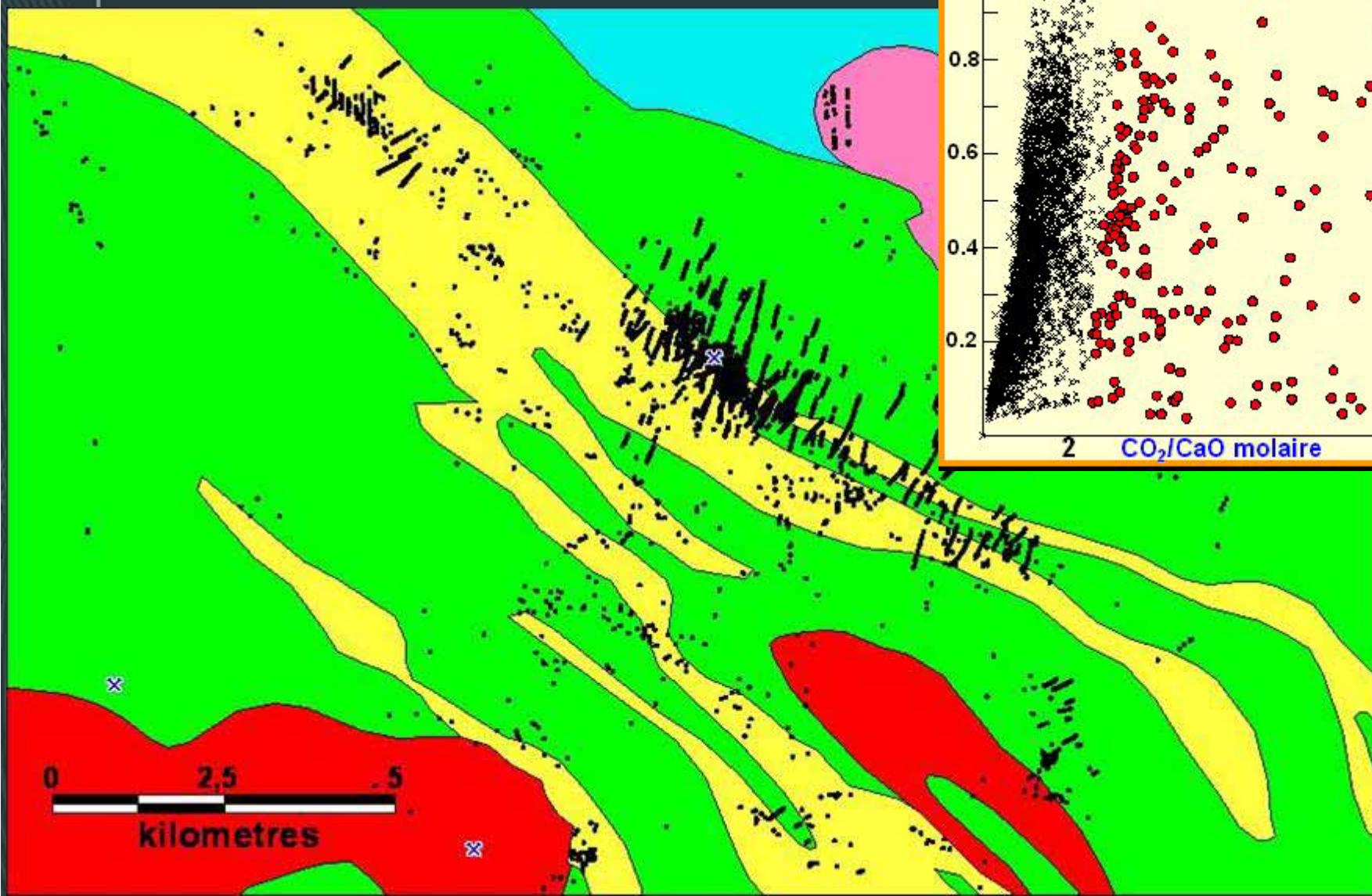


- Échantillons provenant du halo d'altération



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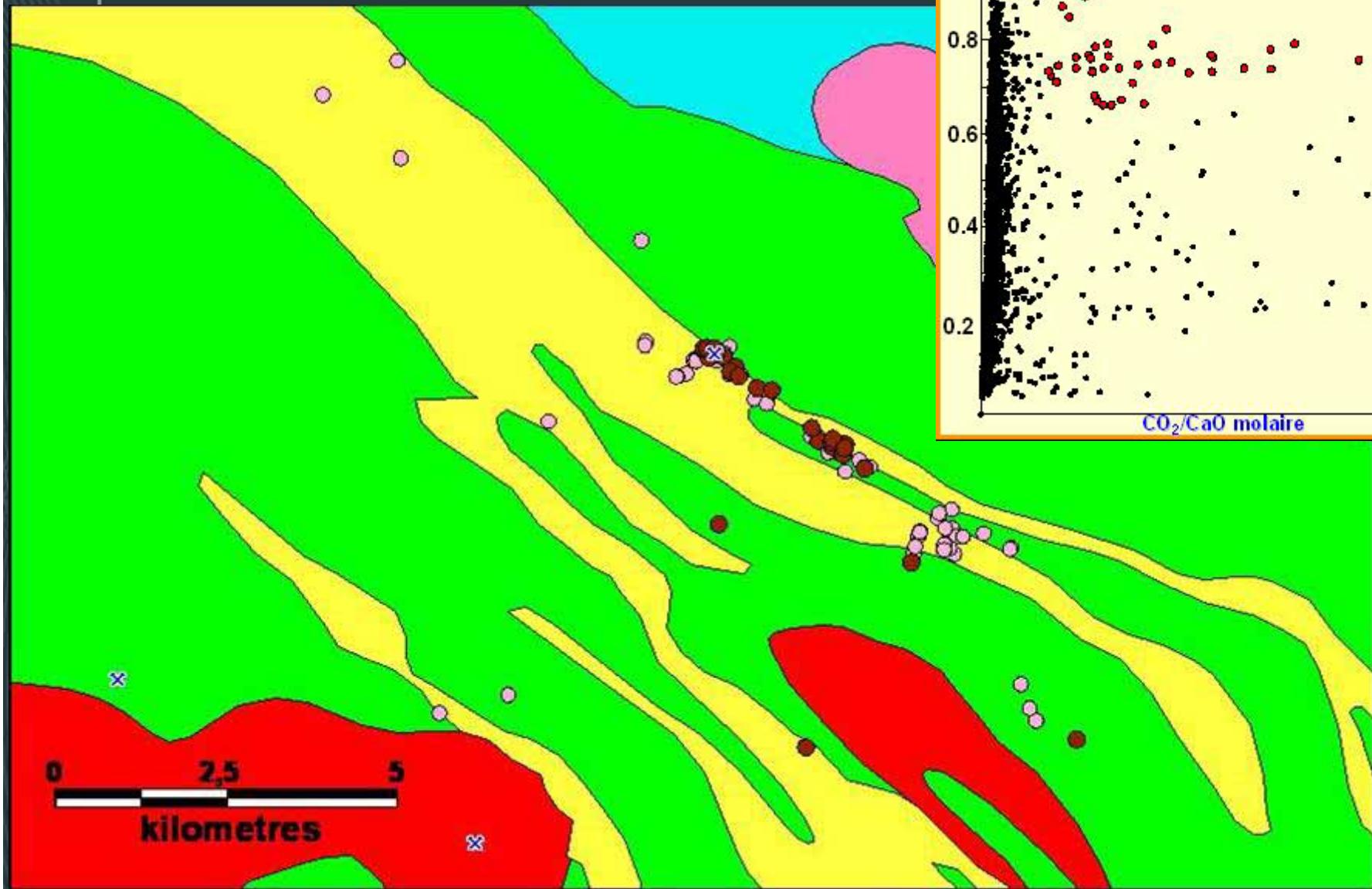
## BOUCHARD HÉBERT





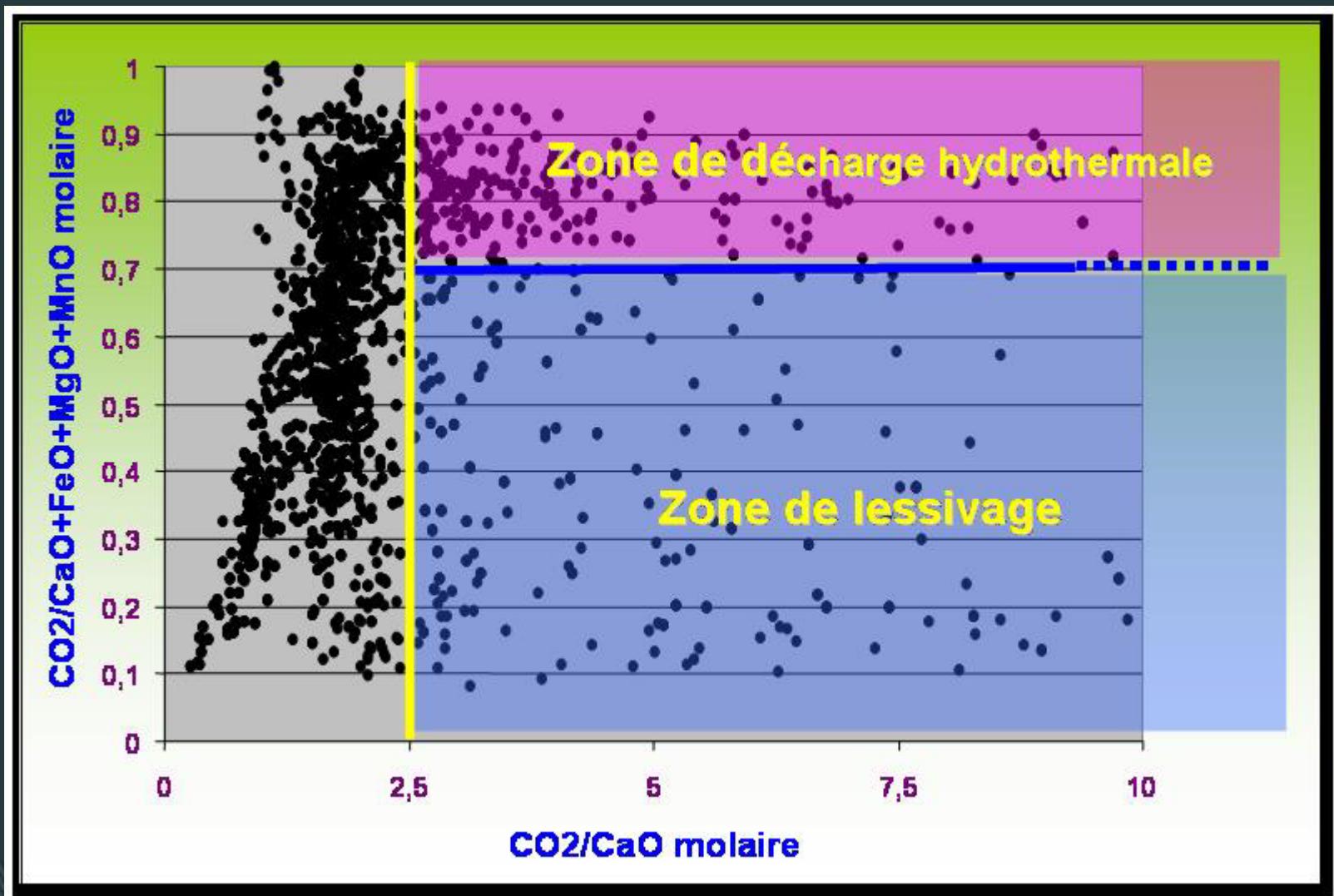
CONSOREM

## BOUCHARD HÉBERT



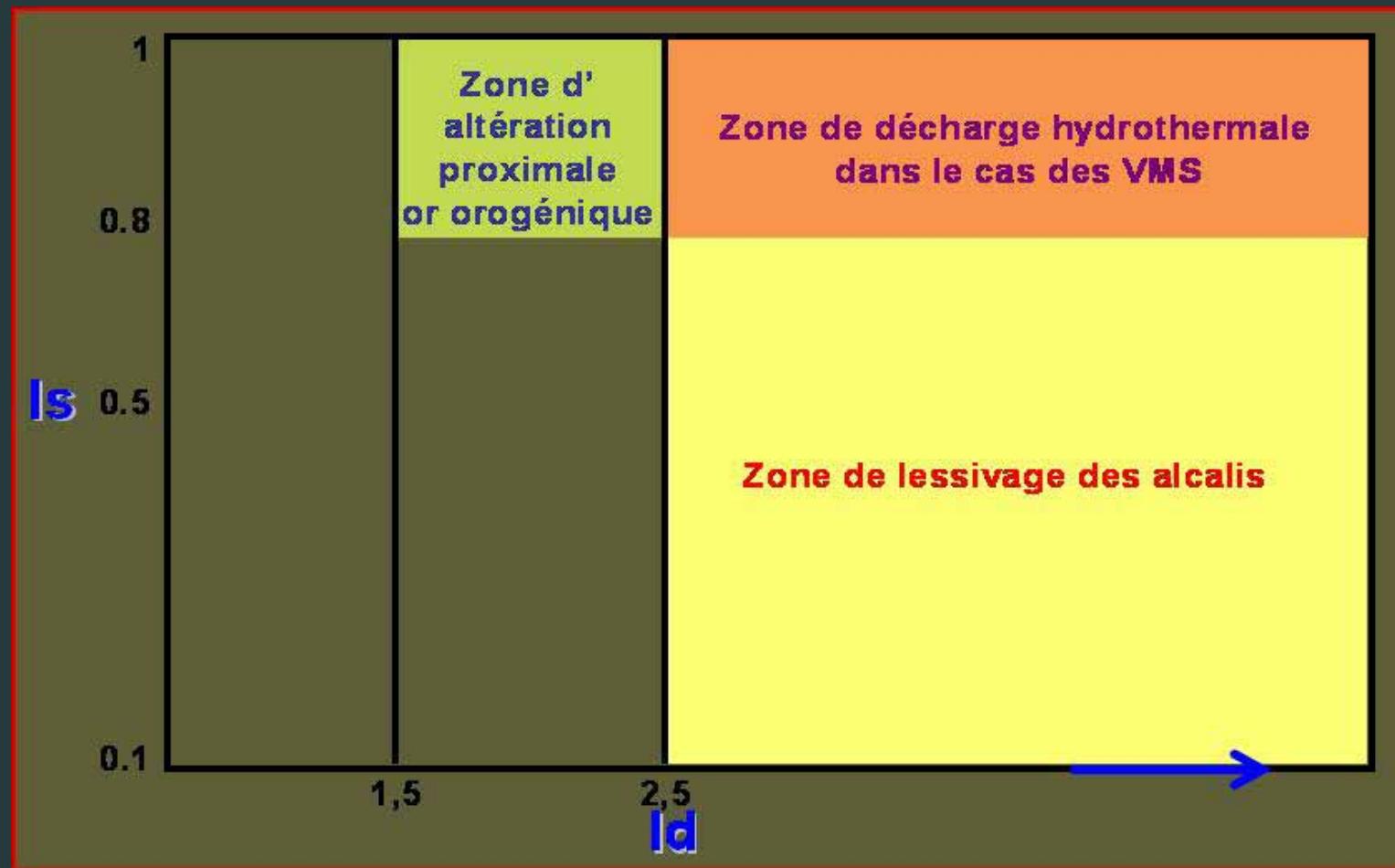


## Diagramme de carbonatation et zones d'altérations proximales





## Diagramme de carbonatation et zones d'altérations proximales





## **Limitations de la méthode**

- Estimation du CO<sub>2</sub>
- Pas de calcul lorsque %CO<sub>2</sub> < 1
- Donne une bonne estimation du contenu total de carbonates pas l'histoire de la carbonatation
- Attention
  - Présence de sulfures dans la roche
  - Présence de veines de carbonates dans la roche



## Conclusions

**Indice de discrimination permet de déterminer les espèces de carbonates à partir d'une analyse lithogéochimique**

**Diagramme de carbonatation est utile pour le ciblage dans les environnements volcanogènes et Au-orogénique.**